



Synthesis and Characterization of "Smart" Biaryl Sultines

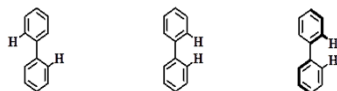
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Rationale

Biphenyl – containing compounds, applicable to many nanoscience applications such as sensors and molecular machines, can be imagined to exist in two states: non-planar or planar. In the non-planar, open state, the π -orbital overlap is minimal, which creates a large dihedral angle between the two rings, thus attenuating electronic communication. In the planar, closed state, π -orbital overlap is substantial, therefore electronic communication between the two rings is facile.



Dihedral Angle = 180 Dihedral Angle = 0 Dihedral Angle = 90

In reality, most biphenyl containing compounds have been known to exist at a dihedral angle of static equilibrium, making these two-states impossible to achieve. However, physical properties such as electron absorption, emission, and conductance are highly dependent on this angle. By synthesizing a reversible tethered sultine bridge, a switching mechanism can be implemented to achieve a two-state switch, and as a result, the fundamental properties of these compounds can be controlled. Thus, the goal of this project is twofold: a) the synthesis and characterization of a donor-acceptor biaryl compound containing a reversible tethered sultine bridge and b) the exploration of dihedral modulation via pH and UV-Vis spectroscopy.

Background

Dihedral Angle:

The angle between two phenyl rings with a single bond as an axis.

Dihedral Angle – Dependent Properties:

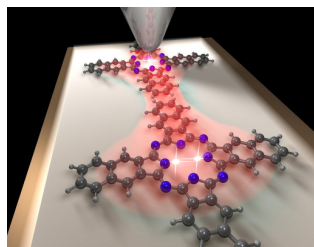
- Absorption wavelength
- Fluorescence
- Electrical Conductance

Molecular Machine:

A discrete functional molecular scale system that is designed to produce quasi-mechanical movements in response to a defined energy stimulus.

Molecular Switch:

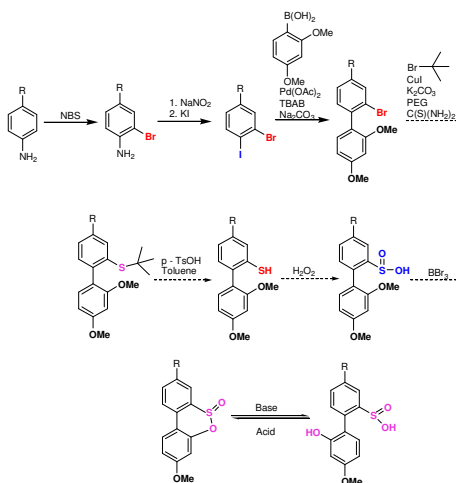
A chemical system, capable of existing in a minimum of two forms, that exhibits different spectral, electrochemical, or magnetic properties. *Interconversion* of each chemical form should only occur as a result of a defined energy stimulus.



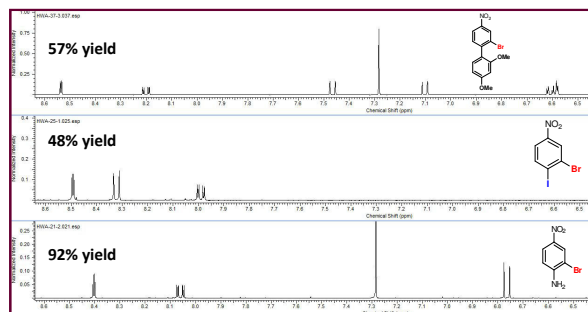
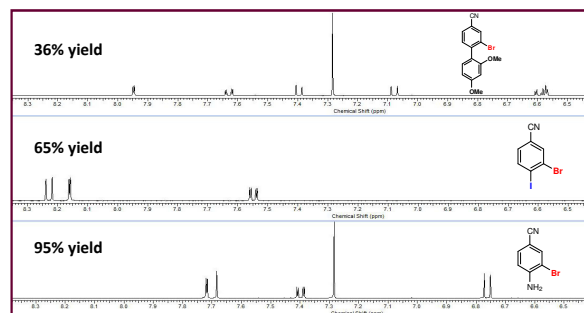
Switching Modes:

- pH environment
- Fluorescence Spectroscopy
- UV-Vis Spectroscopy

Pathway Synthesis

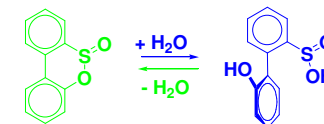


NMR Characterization



Expected Results

pH Reversible Sulfinate Ester Tether



Stable at acidic pH

Stable at basic pH

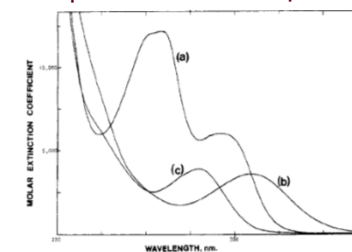
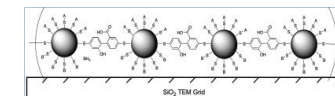


Figure 1. Dependence of the ultraviolet spectrum of biphenylene sultine (2) on pH: (a) biphenylene sultine (2), (b) dipotassium salt of 2-(2-hydroxyphenyl)benzenesulfonic acid (7), (c) 2-(2-hydroxyphenyl)benzenesulfonic acid (8). See Experimental Section for details.

Future Studies

- Synthesis and exploration of alternate novel switches
 - > Terphenyl analogs
 - > Redox switchable biaryls
 - > Light labile bridges
- Conduction of linker containing molecules onto a gold nanoparticle



Acknowledgements

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